

Volume 1

Number 1

Fall 2016

## *The Boller Review:*

---

A TCU Undergraduate Journal of  
Research and Creativity



Jaafari, H.<sup>1,3</sup>, Pendry, R.<sup>1,3</sup>, Nurekeyev, J.<sup>1,3</sup>, Doan, H.<sup>1,3</sup>, Raut, S.<sup>1,3</sup>, Dzyuba, S.<sup>2,3</sup>, Gryczynski, Z.<sup>1,3</sup>  
<sup>1</sup>Department of Physics and Astronomy, <sup>2</sup>Department of Chemistry, Texas Christian University; <sup>3</sup>Center for Fluorescence Technologies and Nanomedicine, University of North Texas Health Science Center

## Introduction

Fluorescence is a phenomenon that provides a means to quantify and characterize how molecules interact with and emit light, particularly molecules of biological nature. Fluorescence lifetime is an important parameter of all fluorescent molecules, and it refers to the length of time a molecule remains in the excited state before emitting a photon; generally speaking, most fluorophores have lifetimes in the range of 0.5 to 20 nanoseconds (ns). However, in order to be practically useful in most biological contexts, fluorescent molecules must have lifetimes around or beyond 20 ns.

Here, we perform fluorescence characterization studies of three closely related fluorescent molecules that are derivatives of boron-dipyrromethene (BODIPY): BODIPY monomer, alkyl-BODIPY, and BODIPY dimer (rotor configuration) (figure 1). The purpose of these characterizations is to identify which molecule has the most optimal lifetime for use in biological applications, such that we can then attempt to hybridize the most optimal molecule with azadioxatriangulenium (ADOTA), a fluorophore that has already proven to be highly useful for large biomolecule binding assays, due to its unusually long lifetime.

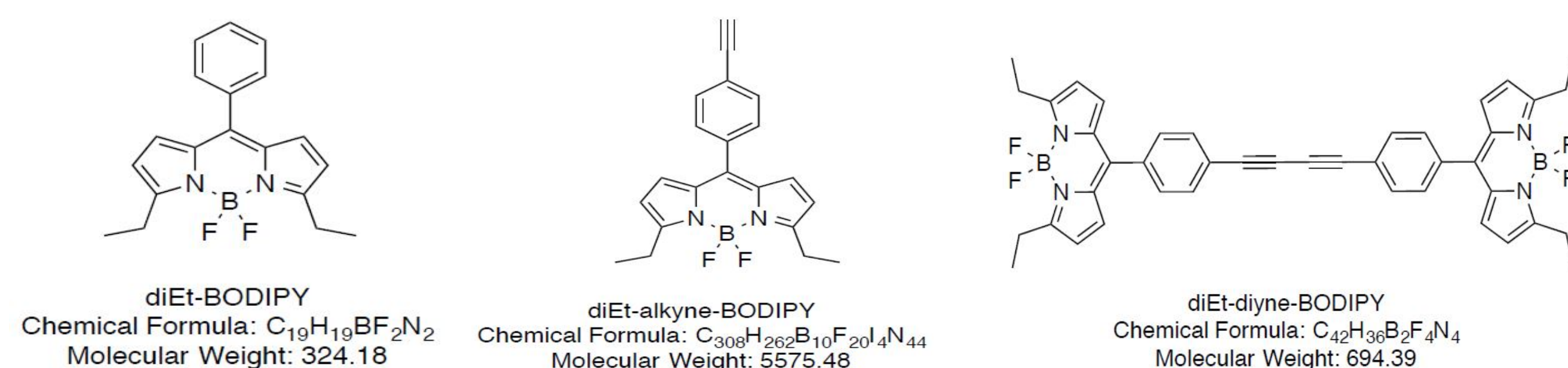


Figure 1: Chemical Structure of BODIPY, alkyl-BODIPY, and BODIPY dimer (rotor configuration)

## Methods

For all three species, we measured fluorescence absorption using linear dichroism and fluorescence excitation/emission anisotropy. Collectively, these data adequately serve to characterize the spectroscopic properties and, therein, the fluorescence lifetime of each molecule.

- In the context of an incident beam, **linear dichroism** refers to the selective absorption of one of the beam's two orthogonal polarization states. Linear dichroism is useful for making polarized absorption measurements of organic dyes within an oriented medium, such as a stretched polymer film.
- Fluorescence anisotropy** is the phenomenon whereby, in the context of fluorescent molecules (fluorophores), the light emitted by a fluorophore will have unequal intensities along different axes of polarization. Consequently, different fluorophores will have different anisotropy values and can be characterized as such. Beyond characterization studies, understanding the anisotropic properties of a fluorophore can tell us a great deal about how the efficacy with which it can bind to a protein and thus be used as a biomarker.

## Data

Within this experiment, we produced polyvinyl alcohol (PVA) films featuring varying concentrations of each dye, and solutions with varying concentrations of each dye. With these solutions and PVA films, we measured the excitation, emission, anisotropy, and fluorescence lifetimes.

### PVA Films

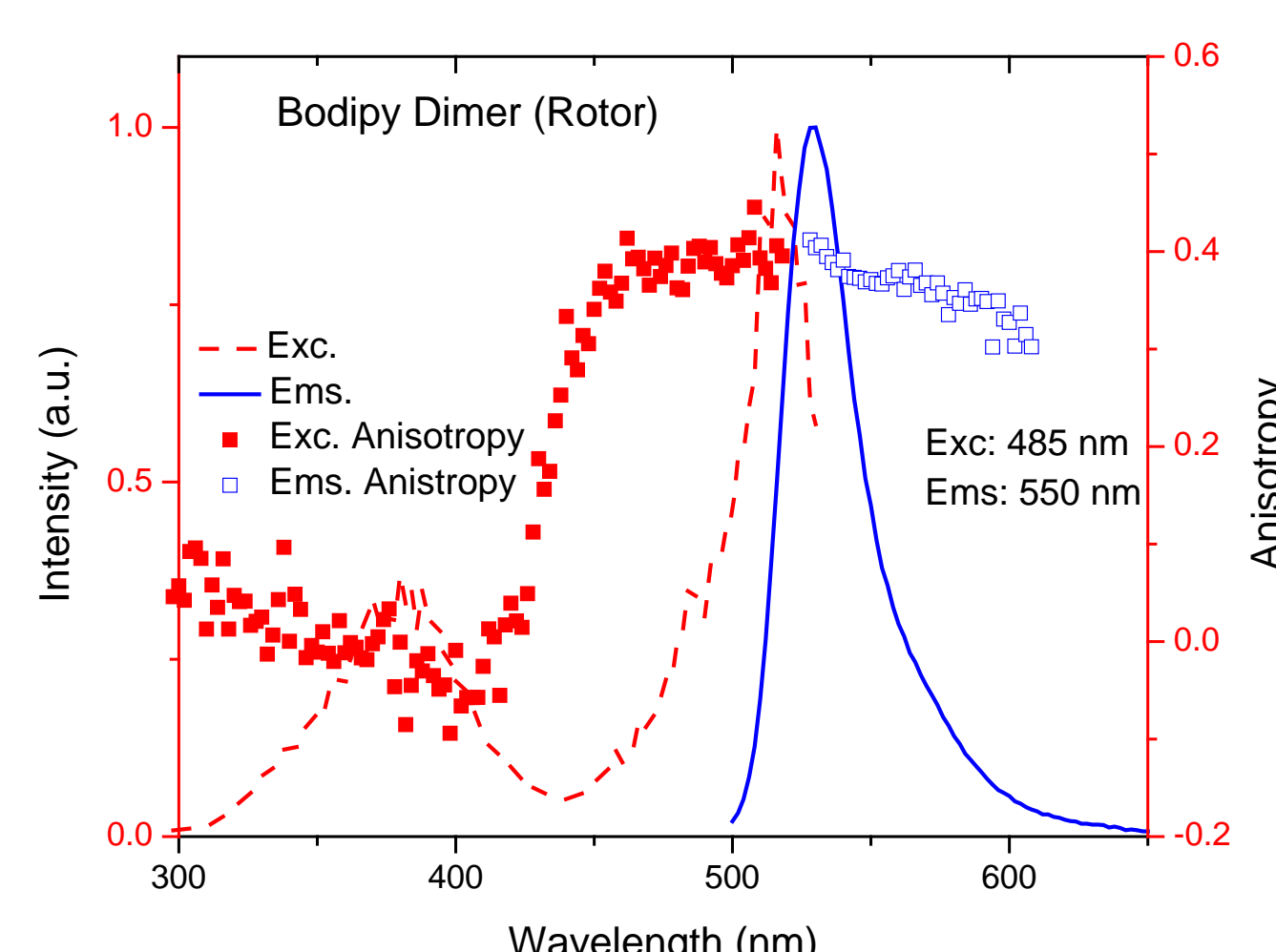


Figure 2. Excitation and emission spectra (obtained via linear dichroism) and excitation/emission anisotropy data for BODIPY dimer (rotor configuration).

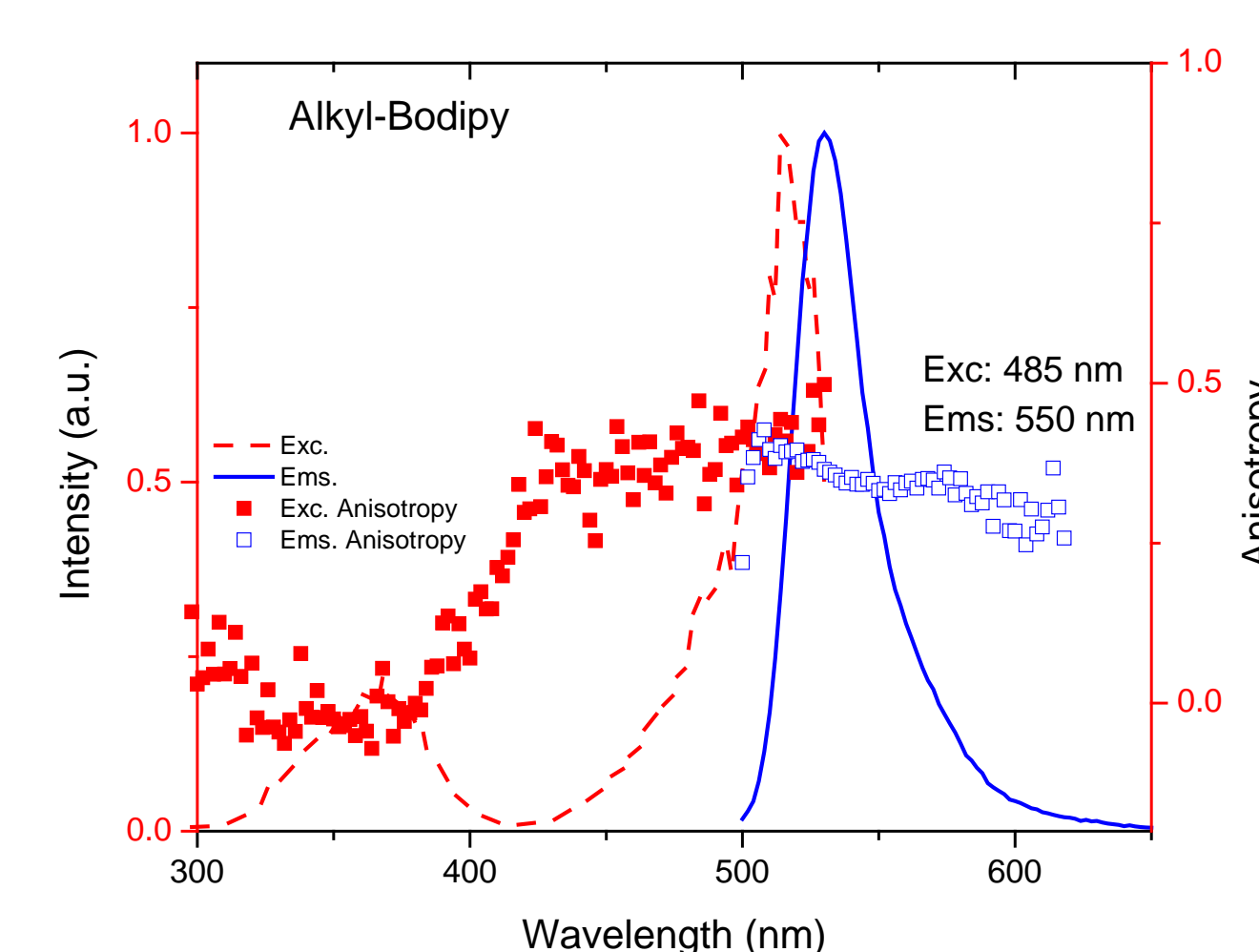


Figure 3. Excitation and emission spectra (obtained via linear dichroism) and excitation/emission anisotropy data for alkyl-BODIPY.

### Solutions

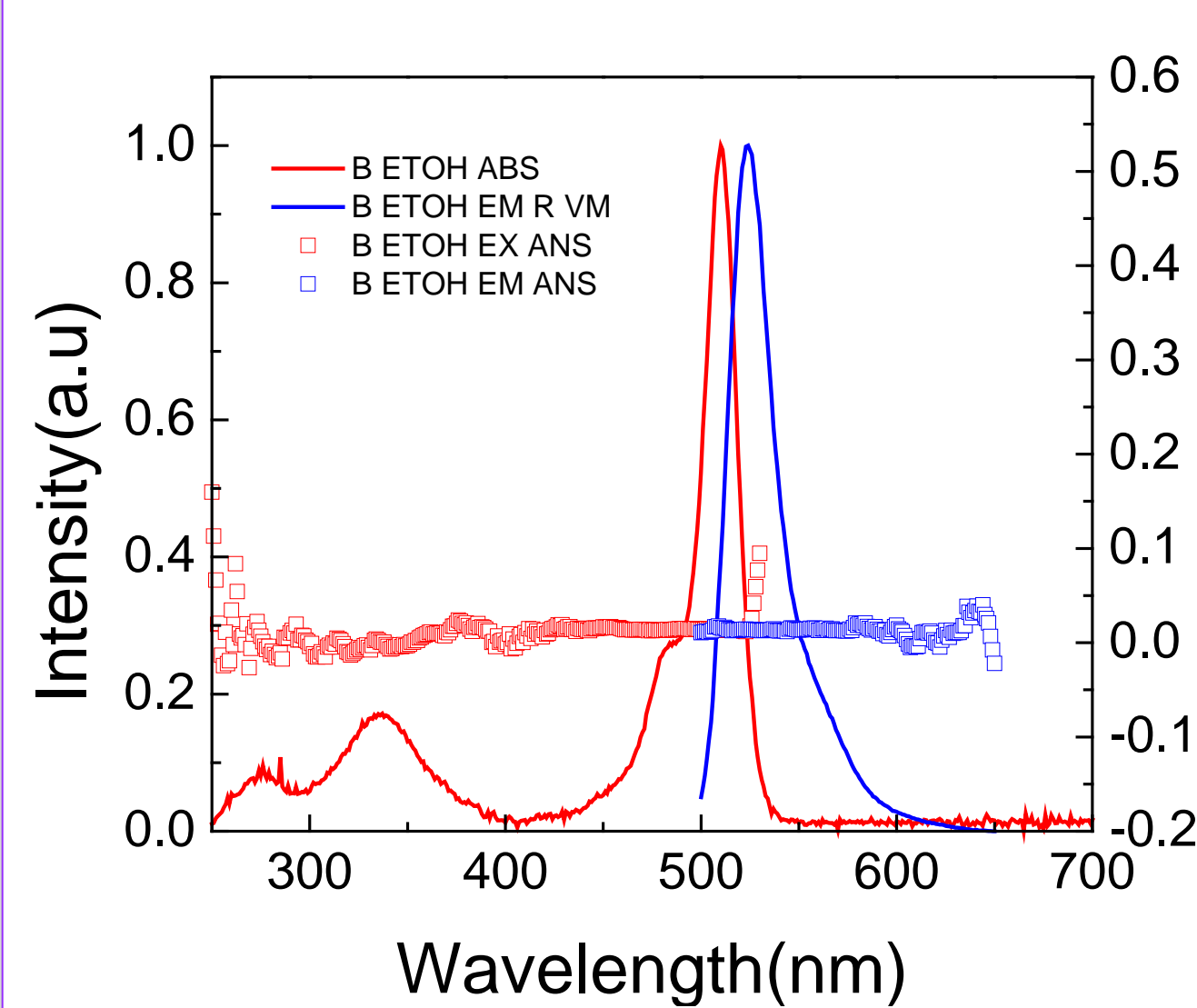


Figure 4. Excitation (ABS) and emission spectra (obtained via linear dichroism) and excitation/emission anisotropy data for BODIPY in ethanol.

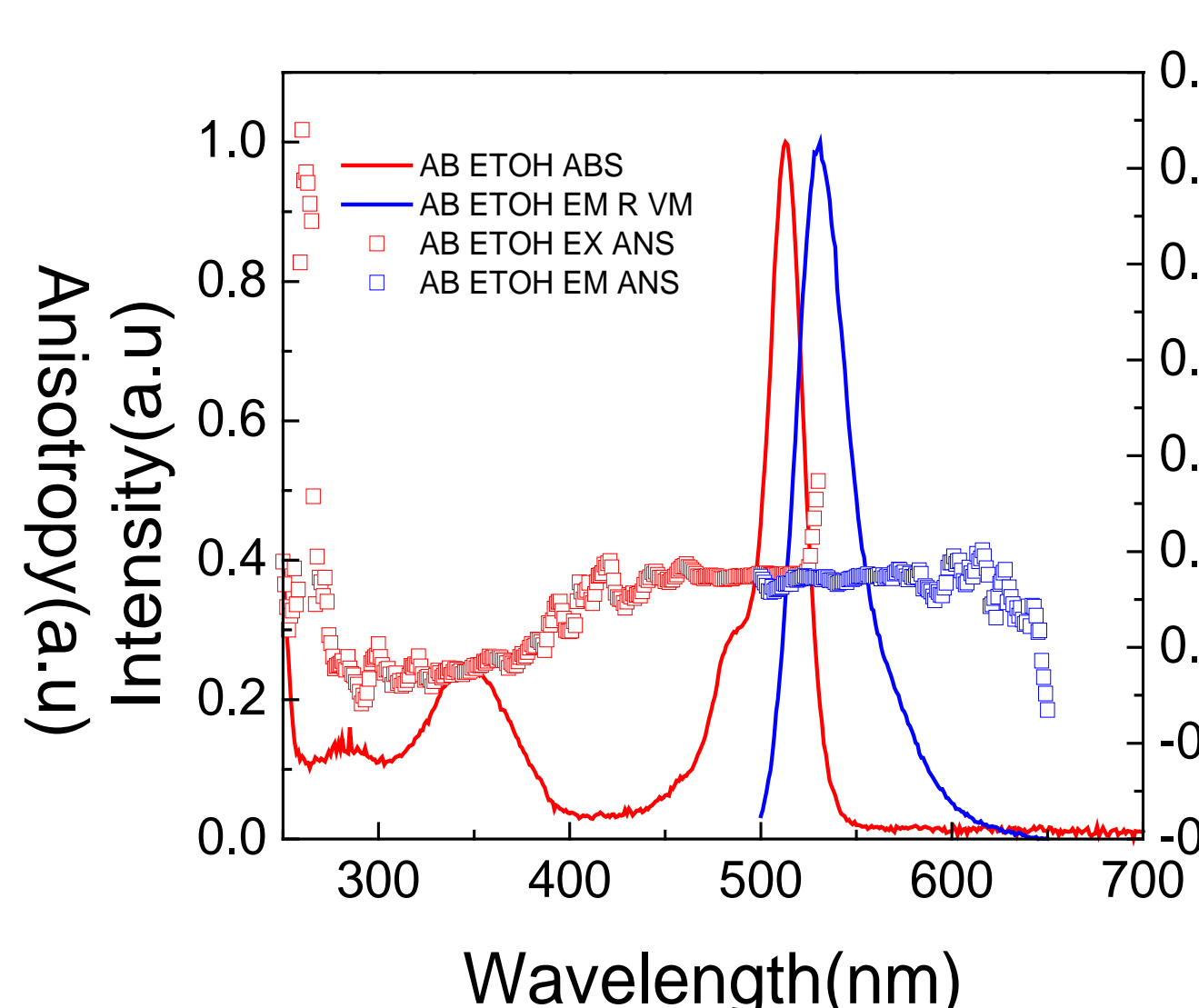


Figure 5. Excitation (ABS) and emission spectra (obtained via linear dichroism) and excitation/emission anisotropy data for alkyl-BODIPY in ethanol.

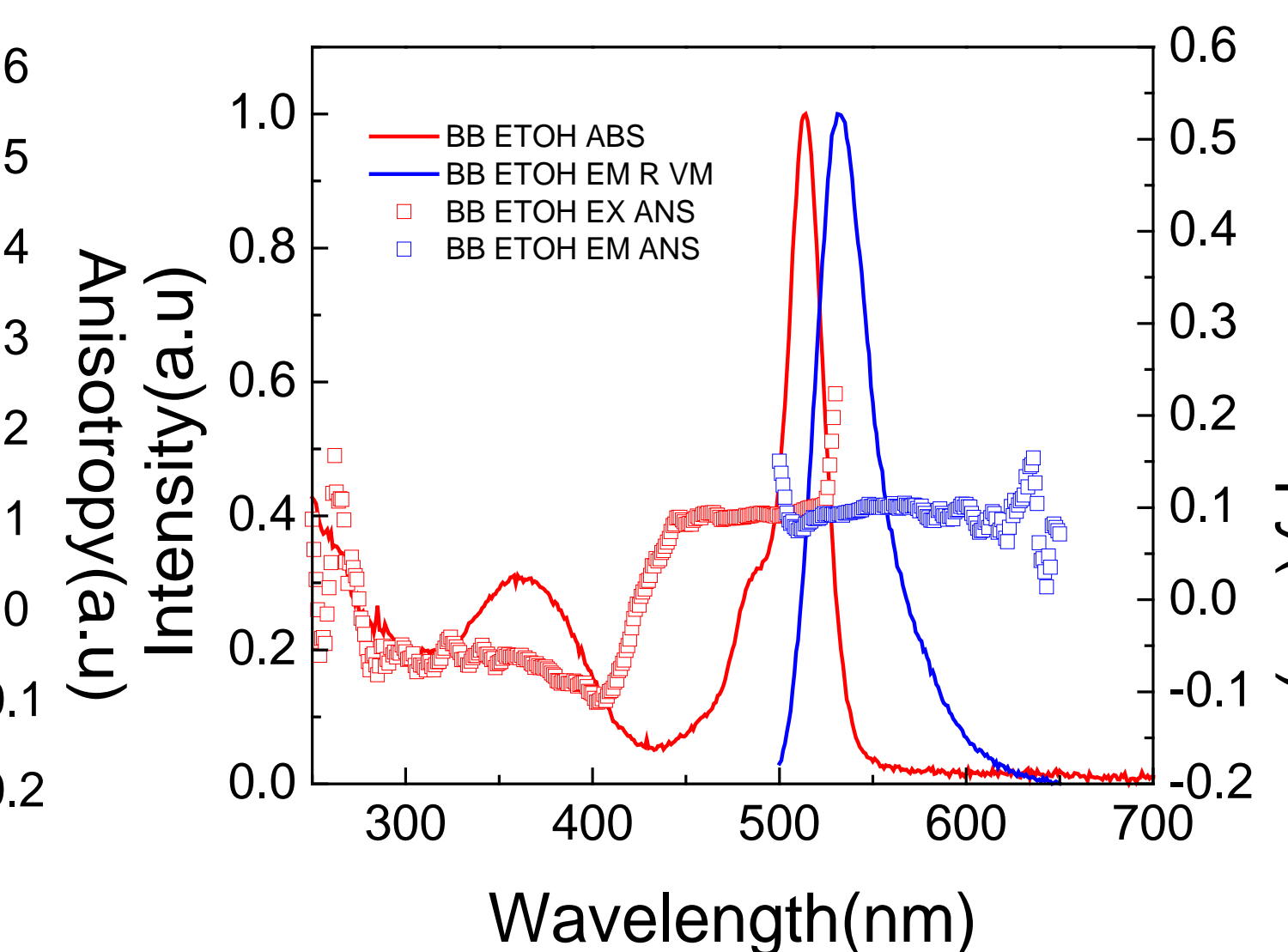


Figure 6. Excitation (ABS) and emission spectra (obtained via linear dichroism) and excitation/emission anisotropy data for BODIPY dimer (rotor configuration) in ethanol.

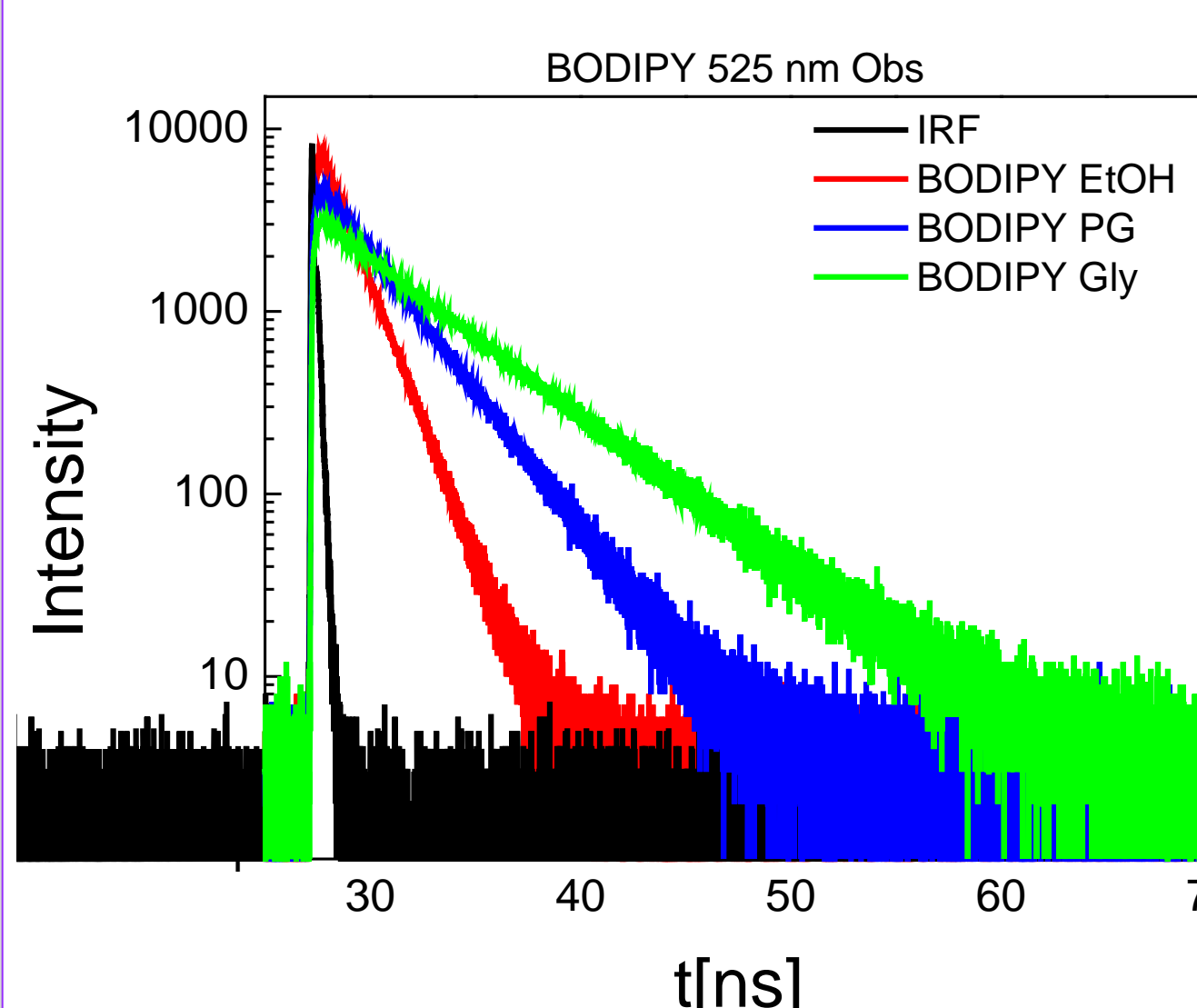


Figure 7. Fluorescence lifetime characterization of BODIPY in solutions of ethanol, propylene glycol, and glycerol.

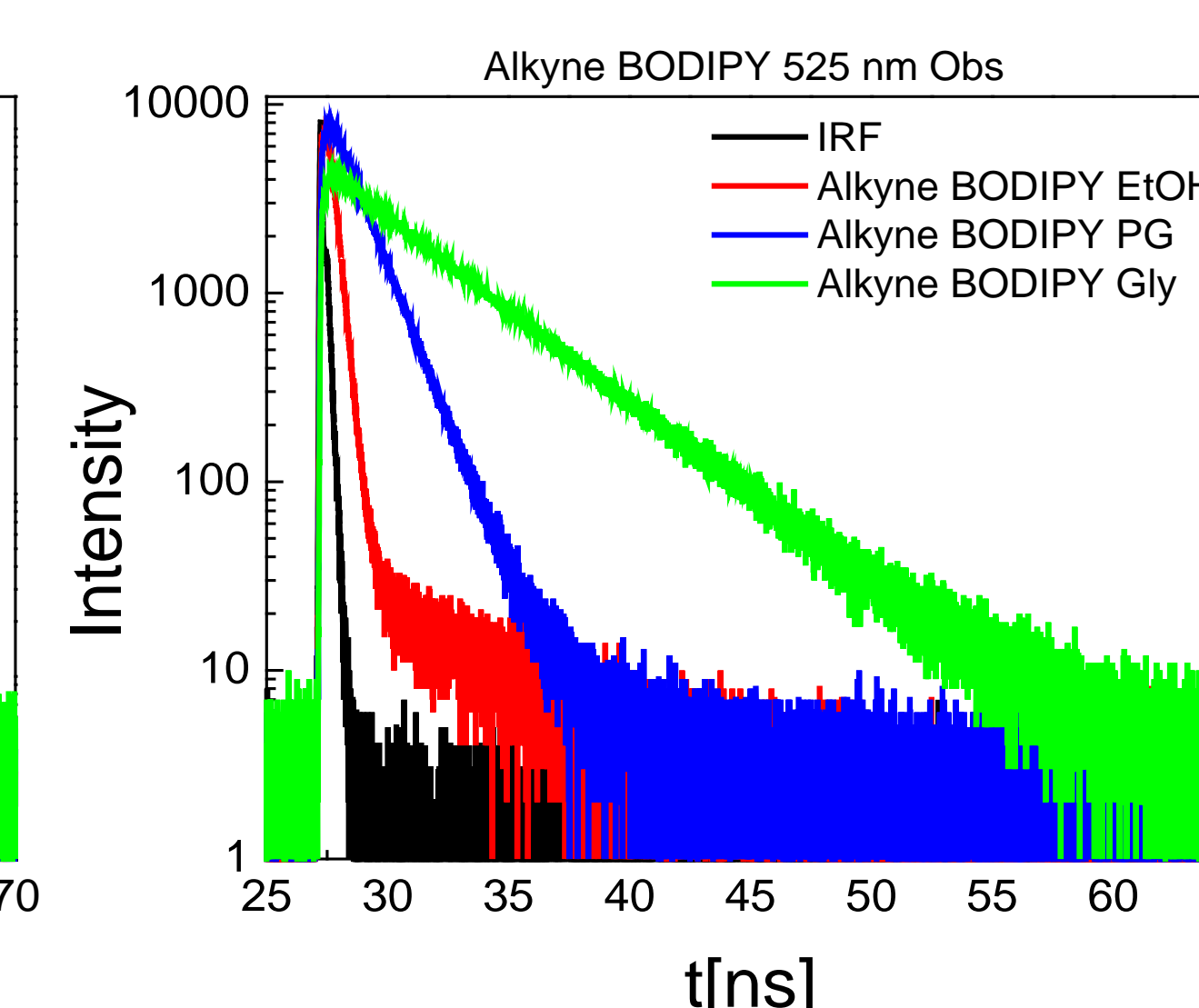


Figure 8. Fluorescence lifetime characterization of alkyl-BODIPY in solutions of ethanol, propylene glycol, and glycerol.

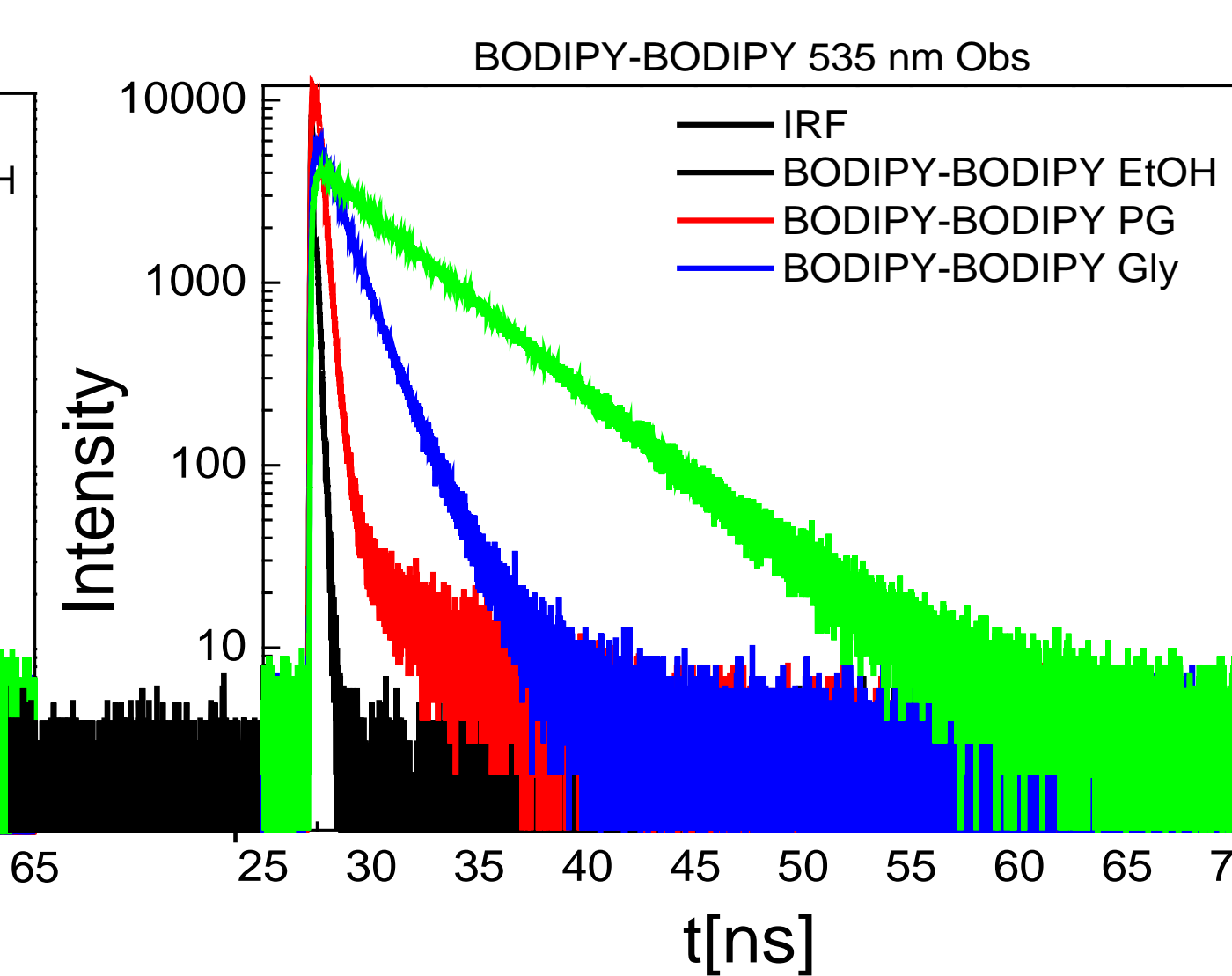


Figure 9. Fluorescence lifetime characterization of BODIPY dimer in solutions of ethanol, propylene glycol, and glycerol.

## Conclusions

Here we performed fluorescence characterization studies of the BODIPY monomer, alkyl-BODIPY, and BODIPY-BODIPY (rotor conformation) via linear dichroism to measure excitation and emission properties, as well as via anisotropy measurements. Of the three species studies, BODIPY-BODIPY had the most optimal lifetime measurements, especially in glycerol solution, which is most akin to the environment that would be found in the extracellular milieu or a typical cell cytosol. The lifetime of BODIPY-BODIPY in polyethylene glycol and in ethanol was comparable to the lifetime of BODIPY and alkyl-BODIPY in the same solutions. In stretched PVA film, the anisotropy measurement for BODIPY-BODIPY was approximately half that of alkyl-BODIPY; being that anisotropy is a measurement of the ratio of polarized photons emitted to total photons emitted, a lower ratio is indicative of less interference and higher quantum yield, implying that BODIPY-BODIPY is the most optimal of the three dyes.

## Future Directions

The purpose of these characterization studies was to determine the most practically useful BODIPY species for use in a biological setting, where dyes with long lifetimes are especially useful. With our results indicating BODIPY-BODIPY as the most optimal species of fluorescent dye, we now aim to (in collaboration with Dr. Sergei Dzyuba of the TCU Chemistry Department) hybridize this dimer with ADOTA (figure 10), a fluorescent molecule with an abnormally long lifetime, to determine if their lifetimes synergize to make an ultra long-lived fluorophore that would thus be both highly optimal and intriguing for studies in a biological and pathological contexts.

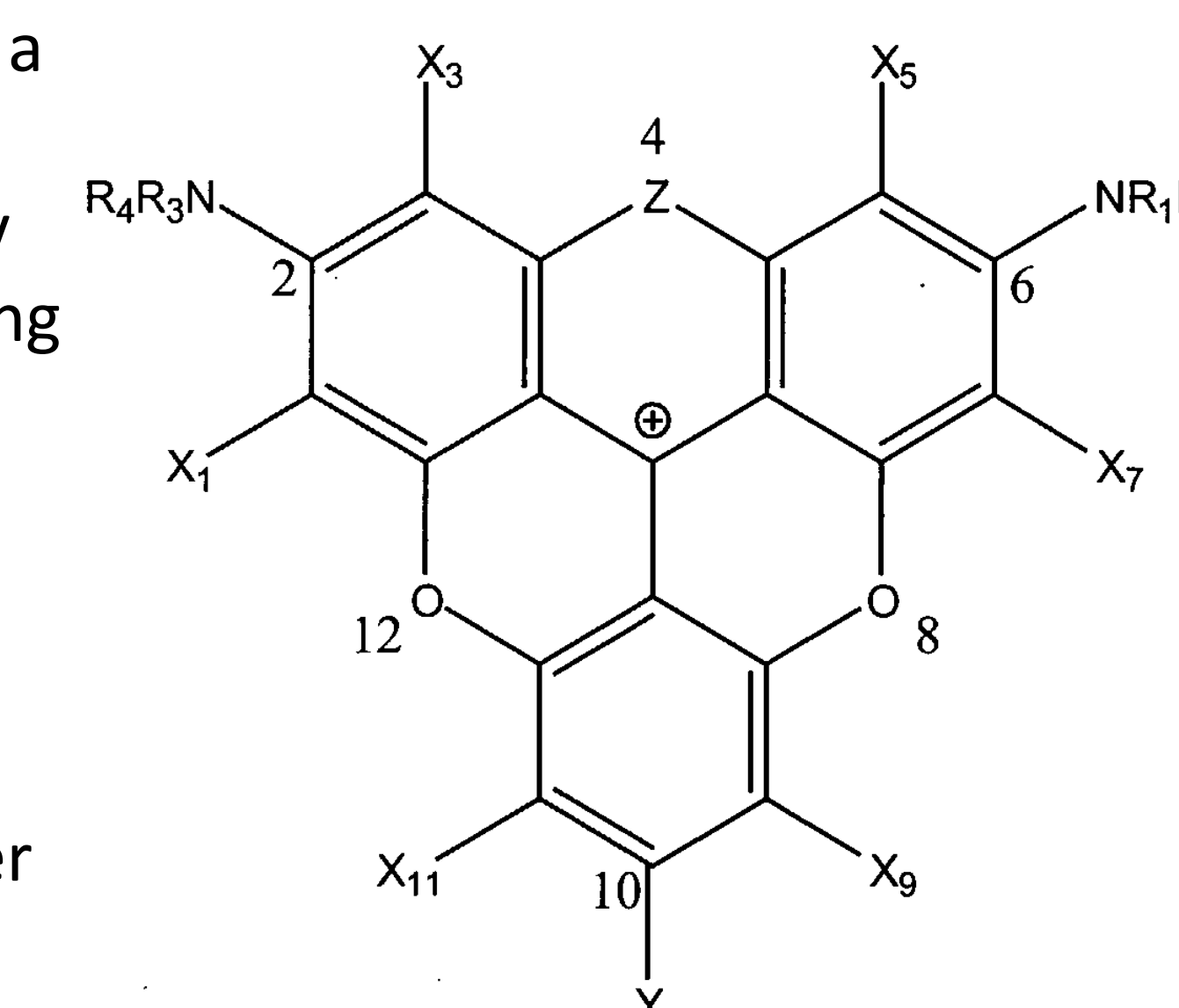


Figure 10. Structure of the azadioxatriangulenium (ADOTA) fluorophore.

## Acknowledgements